EFFECTS OF NANO-CERIA BASED FUEL BORNE CATALYST ON SINGLE CYLINDER DIESEL ENGINE CHARACTERISTICS NGHIÊN CỦU ẢNH HƯỚNG CỦA XÚC TÁC NHIÊN LIỆU ÔXÍT NANO XÊRI ĐÊN ĐẶC TÍNH CỦA ĐÔNG CƠ DIESEL I XILANH

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ABSTRACT

Aiming to reduce the fuel consumption and exhaust emissions of Internal combustion engines, nano-cerie (CeO₂) based Fuel Borne Catalyst (nFBC) developed by NanoScience Innovation Pte Ld, Singapore has been directly blended with Vielnam's market diesel fuel in the ratio of 1:5,000. This blended diesel consists of more than 99.99% diesel, less than 0.01% surfactant (hydrocarbon), and less than 10pm (10mg/tite) of CeO₂ nanoparticlas.

Comparative experiments were conducted on single cylinder research engine AVL 5402 to find the impacts of this nFBC additive on engine's performance, exhaust emissions and combustion profile. The experimental results show that there are no adverse effects on the engine after 56-hours running with nFBC. The reduction of fuel consumption increases gradually and reaches 7.7% after 55-hours running with nFBC additive. The soot concentration has declined on the averaged of 10-20%. The carbon monoxide (CO) emission has increased significantly in the early engine running stage (less than 20-hours) which might be due to the burning of the deposited carbon through catalytic cleaning effect. However, after 20-hours running with nFBC additive, the CO emission has decreased dramatically. The total hydro-carbon (THC) emission has declinesed by more than 10% in various operating regimes. However, nitrogen oxides (NO₂) emission reduction has not been as significant as expected. The peak of combustion pressure profile has moved closer to firing top dead center at the positive crants angle regime after adding nfBC Into the fuel. This phenomenon has been baseved most prominent at low engine speed (1,400rpm). However, at higher engine speeds, the time point of pressure peeds has not changed significantly.

Keywords: Nano fuel's additive, combustion characteristics, single cylinder diesel engine.

ΤόΜ ΤΑΤ

Hướng tới giảm tiêu thụ nhiên liệu và phát thải độc hại của động cơ đốt trong, xú tác nhiên liệu dựa trên cơ sở ôxilt xêri dạng nano (nFBC), phát triển bởi Công tr NanoScience Innovation, Singapore được trận với nhiên liệu diesei thị trường ở Việt Nam với tỷ 81:5000. Hồn hợp điesei này bao gồm trên 99,99% diesei, dưới 0,01% chất bề mặt (hyđró cácbon) và đưới 10ppm (10mg/lit) hẹt ôxilt xêri dạng nano.

Các thừ nghiêm đối chứng được thực hiện trên động cơ tiêu chuẩn AVL 5402 nhằm đánh giễ ảnh hưởng của phụ gia n FBC đến tinn năng, phát thải và diễn biến áp suất trong xilanh đặng cơ Các; kết quả thực nghiệm cho thấy, động cơ vận hành ôn định với phụ gia n FBC; tiểu thụ nhiên liệu giảm dân và đạt 7,7% sau 56 giờ vận hành; độ mở khỏi giảm trung bình 10-20%; phát thải mônôxit cácbar, (CO) chi giảm sau 20 giờ vận hành; trước thời giảm trận QC tăng mẹnh do hiệu ứng tàm sech uôngi chây của chất xúc tác; phát thải hyđrô cácbon tông giảm trên 10% đối với hảu hất các chế độ vận hành. Tuy nhiên, phát thải NO, không được cải thiện như dự tính. Thời điểm áp suốt cự đại trong lành được ciộc hcuyển theo chiều hưởng gần với điểm chất trên trong hành trình giãn nở khi có mặt phụ gia nFBC. Hiện tương này được thể hiện rõ ở tốc độ vòng quay thắp của đội cơi hấng cơ (1400 vịnh), trự nhiên, ở tốc độ quay cao, thời điểm áp suất cực đại không có nhiều thay đổi khi có sử dụng phụ gia nFBC.

Từ khóa: phụ gia nano cho nhiên liệu, đặc tính cháy, động cơ diesel 1 xilanh.

1. INTRODUCTION

Cerium Oxide, CeO2, has been known for its redox (reduction and oxidation) catalytic effects in various internal combustion engines and has been studied over the last decade When nano-ceria (CeO₂) based Fuel Borne Catalyst (nFBC) is mixed in fuel at very low concentration, they can oxidize carbon-soot (Csoot) and total hydrocarbons (THC) at lower temperatures to become CO2 and H2O. Furthermore, C-soot attached to engine's wall can also be removed through burning due to the catalytic activities of CeO₂, allowing the engine to work more efficiently. Very little amount (5-10ppm) of nFBC with size of 5-10nm would have sufficient surface area for catalytic effect to reduce fuel consumption and all types of emission from the internal combustion engine. This series of tests is done on diesel internal combustion engine. This type of fuel additive is considered as one of the most advanced nanotechnology additive available which is being developed, tested and used in several countries like UK, USA, Singapore, and Australia etc [1].

Figure I shows the particulates distribution from a standard diesel engine exhaust. It is observed that significant amount of particulates is in nano-scale (<50nm) [2]. So the additional of 10pm nFBC would not significantly change the nanoparticles population. However, its effect on reducing these carbon based particle is significant.



Fig. 1 The size distribution of typical diesel exhaust particles ranges from nano-scale particles up to coarse soot particles [2].

2. BACKGROUND OF CEO₂ CATALYTIC EFFECTS

It is beneficial to discuss the roles that η FBC plays in each reaction which may influence the fuel consumption, CO₂, CO, NO₃, THC emissions and opacity separately and concurrently.

It is well known that CeO₂ nanoparticles provide the catalytic activities on oxidation and reduction which can be expressed in the following reaction equations [3],[7]:

$2CcO_2$	$<=> Ce_2O_3 + V_0^{\bullet} + O^{\bullet}$	(1)
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 $(2x+y)C_{0}O_{2} + C_{x}H_{y} \Rightarrow [(2x+y)/2]C_{2}O_{3} + xCO + y/2H_{2}O$ (2)

$$2CeO_2 + C_{yoot} \Rightarrow Ce_2O_3 + CO$$
 (3)

$$4CeO_2 + C_{soot} => 2Ce_2O_3 + CO_2$$
 (4)

$$2CeO_2 + CO => Ce_2O_3 + CO_2$$
 (5)

$$Cc_2O_3 + NO_x \Longrightarrow Cc_2O_{3+x} + 1/2N_2$$
 (6)

Eq-1 describes oxygen storing/releasing effect of CeO₂ nanoparticles These oxygen (O°) are un atomic form, reside on surface of CeO₂ nanoparticles and is more active than molecular oxygen in air (O₂). As a result, O^o can reduce the temperature of oxidation reaction in Eqs-2, 3, 4, and 5 to as low as 300°C. Eq-6 shows the reduction process of NO₄ to N₂. However, NO₄ can be also formed through combining N₂ and oxygen from either air or O⁶ from CeO₄ at levated temperatures.

Under normal circumstances, the engine wall temperature is lower than the carbon combustion temperature, part of the unburnt carbon would deposit onto wall when the engine runs. This wall deposit would slowly accumulate leading to engine performance degradation. With the presence of CeO₂ in the system, the catalytic effect would lower the carbon combustion temperature which leads to burn-off the carbon deposit on the wall. In other words, the CeO₂ nanoparticles would deposit on and clean the wall. The CeO₂ deposition would reach an equilibrium amount and the engine is then expected to run in more efficient mode.

From the processes described above, detailed analysis of combustion process

involving CeO₂ becomes very complex as the CeO₂ particles can either participate in oxidation or reduction reaction. Concentration of the various exhaust emissions such as CO, C-soot, THC and NO₄ are expected to fluctuate widely, especially in the early stage of the test. The stability of the exhaust can only be established after relatively long hours of operation until some kind of equilibrium between the surfaces of combustion chamber and the combustion process. This is one of the purposes to make 3 different time frame in this test.

In general, CO, C-soot, THC can be classified in a same category and oxidation reaction is favorable in terms of energy oxidation of these gases to become CO₂ is one way reaction (releasing energy) at all working temperatures. In other words, by adding the CeO₂ nanoparticles, these emissions will decrease eventually and they occur either inside or outside the combustion chamber.

3. EXPERIMENTAL

The purpose of the test was to systematically investigate the time history on the effect of ηFBC on diesel engine combustion. It is expected that the engine will run more efficient and discharge less pollutant gases. Adversary effects, if any, to the engine due to the additive will also be reported in this long duration of running of the engine.

3.1. Test procedure

In the beginning, the engine was fueled and run with the diesel acquired from Vietnamese open market. Finheen different engine running modes with various loads and speeds of 1,400 rpm, 1,800 rpm and 3,000 rpm were carried out. The fuel consumption and exhaust emissions were measured and recorded for reference. Straight after that, the same diesel was added with 5-10 ppm of η FBC and another cycle of the same 15 running-mode was carried out on the same engine. Similarly, the fuel consumption and the exhaust emissions were measured and recorded for comparing with test running with and without additive.

Engine was then run for 20 hours continuously with ηFBC added fuel at

1,800rpm and 50% load before the third 15 running-mode cycle test and measurement were carried out.

The final 15 running-mode cycle test was performed after 56 hours of engine running with nFBC at 1,800rpm and 50% load.

The data of fuel consumption and exhaust emissions at different speeds and loads varied widely. Although running parameters are not exactly the same as standard, it is useful to average over a standard European ECE R49 protocol. Figure 2 shows the engine's operating modes and the weighing factor for each mode of ECE R49 protocol.



Fig. 2 European ECE R49 cycle [1].

3.2. Test apparatus

Test has been done on one cylinder Austrian made research engine - AVL 5402. This test bench is controlled by a computer system with different modules like PUMA, EMCON (6). The electronic brake system -AMK, can work at maximum power, torque and speed at 28kW, 150Nm and 7,000pm respectively. Exhaust emissions were analyzed by CBBII work bench, which is able to monitor different gases like CO₂, CO, NO, NO, and THC [5]. The control system of the test set up is presented in Figure 3 and the configuration of the test engine is presented in table 1.

Items	Values	Units
Bore	85	mm
Stroke	90	mm
Displacement	510.7	cc
Compression ratio	17:1	-
Rated power/speed	9/3200	kW/rpm

Table 1 Specification of the test engine

AV1.5402



Fig. 3 Control system of the experimental set up.

4. RESULTS AND DISCUSSIONS

4.1. Pressure profiles

Figure 4 shows the combustion pressure profiles for running with and without ηFBC at cagine speed of 1,400pm. The pressure profiles have changed significantly when ηFBC was added into the fuel. The combustion pressure has increased and the pressure front moved forward closer to 0 degree crank angle.

It is believed that nFBC have catalyzed the ignition of diesel earlier. The pressure increment occurs at the positive crank angle regime where actual useful mechanical work is done. In this circumstance, the nFBC have promoted combustion within diesed droplets in fuel rich environment. Due to their ability to initiate the combustion at lower temperature, more fuel would be combusted.

On the other hand, for the case of 1,800rpm and 3,000rpm, the changes are not significant although in both cases, higher

pressures also move toward positive crank angles. This may be due to both temperature and residence time in favor for catalytic effects of nFBC for the low rpm (1,400rpm) operation for this particular test engine.



Fig. 4 Pressure profile versus crank angle at 1,400rpm with nFBC (PCYL1-Diesel w.FBC) and without nFBC (PCYL1-Diesel).

4.2. Fuel consumption

The measured fuel consumption was weighed-average fuel according to ECE R49 protocol and shown in Figure 5. This figure clearly shows the fuel consumption has reduced over time. It is observed that, after 56 hrs nunning with ηFBC, the fuel consumption improvement reaches 7.7%.



Fig. 5 Fuel improvement versus running time with nFBC.

The fuel consumption improvement could be considered coming from two fronts, i.e. more complete combustion due to oxidation occurring at lower temperature and the assumption on engine cleaning effect.

4.3. Exhaust emissions

Total hydrocarbon emission was measured through heated ionization flame detector and the measured and weighed average results are shown in Figure 6. Overall, the THC improvement is higher than 10%.



Fig. 6 THC emission improvement versus engine running time with nFBC.



Fig. 8 NOx emission versus engine running time with ηFBC .

The measured and weighed averaged results of CO emission are shown in Figure 7. Comparing the emission between with and without nFBC, the results fluctuate widely, from worsen to much improved. Immediately after the nFBC were added to the fuel (0th hour), the CO emission worsen which might be due to engine chamber cleaning cycle. The CO emission reduces dramatically by 22.5% at 20th hour of engine running with the additive. However, the improvement of CO emission reduces after 56 hours of engine running.

CO emission come from oxidation process as indicated in Eq-2 and 5. The CeO₂ especially the carbon deposit on the engine wall. Therefore, it is logical to expect the increase of CO emission in the early stage of engine running. However, the engine will slowly reach its equilibrium state at which CO emission will decrease due to catalytic effects of ηFBC over time.

On the other hand, the CO_2 will not be expected to decline further even if the engine operates more efficiently as final products of all oxidation reactions lead to CO_2 .

Figure 8 shows the weighed average of NOx emission at different stages. The emission



Fig. 7 CO emission improvement versus engine running time with ηFBC .



Fig. 9 Opacity index improvement versus engine running time with nFBC.

also fluctuates over time but small percentage of NO_x reduction has been measured after 56 hour engine running.

As discussed on mechanisms of the NO. formation, it is very difficult to pin-point which mechanism is dominated at the present situation. Theoretically, combustion with nFBC would lower the temperature and hence the NO. formation through 'thermal-NO.' mechanism would decrease. On the other hand the presence of O* at the nanoparticles surface would increase the 'prompt- NO_a' formation. 'fuel-NO,' mechanism is normally The important only for biodicsel due to its nitrogen content in the fuel which is not quite relevant here. These competing mechanisms might play different roles at different stages of the test.

When correlate to the CO emission data, the CO emission reduction has caused the NO, emission to increase. This can be interpreted as that the temperature inside the combustion chamber has increased. Nevertheless, in overall, all emissions have improved over time.

In addition, after the gases discharge from the combustion chamber, there is another opportunity of reducing NO_x to N₂ through reaction described in Eq-6 when the CeO₂ is present. This reaction is commonly used in the combustion-engine's catalytic converter.

Figure 9 shows the averaged opacity index improvement of the engine at different running stages. The improvement was found to be more than10%.

5. SUMMARY AND OUTLOOKS

In summary, no adverse effects were encountered with the engine running with the diesel fuel blended with nFBC.

The test has shown the reduction of fuel consumption and gases emission of the diese engine when PFBC (5-10ppm) are added into diesel fuel. Fuel consumption reduction has eventually reached 7.7% after 56 hour running with nFBC.

The THC emission has overall declined under using of η FBC. The CO emission has increased in the early stage but after running for a longer duration, it has decreased to below the level of that in case without η FBC. Although the NO_x emission has yielded mixed results it seems to be correlated to CO emission. The increment of NO_x emission corresponds to high CO reduction which may due to some changes of the in combustion environment during that stage. However, after 56 hours of utilizing η FBC, the engine emission level is lower than the emission without η FBC.

Opacity index, an indirect measurement of particulate matter emission has also shown more than 10% of reduction. Chamber pressure profiles have also shown higher combustion pressure at the positive side of crank angle, leading toward more positive work done

From the emission fluctuation level, it seems that the engine still has not reached its equilibrium state after 56 hours of running with the additive. It is expected to take sometimes for the combustion chamber to be 'cleaned' through canalization of the deposited carbon.

Once the equilibrium amount of η FBC have deposited onto the engine's wall, this layer of η FBC would assist not only in removing carbon to be deposited on the wall, but also catalyzing carbon and THC burn-off at as low as 300°C. This will lead to more completed combustion inside the combustion chamber and hence, better fuel efficiency. A further detailed study of this effect will be very useful.

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